LOHC dehydrogenation catalyst synthesis by ball milling

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Liquid organic hydrogen carriers (LOHCs) provide a promising form of hydrogen storage to enable green hydrogen usage as a carbon neutral energy carrier. The hydrogenation/dehydrogenation reactions of LOHCs are reversible and effective when suitable catalysts are used. However, the dehydrogenation reaction is industrially a new process, and the catalysts are still under development. Supported platinum catalysts are highly active in the dehydrogenation reaction, but they suffer from deactivation. Ball milling (BM) is an emerging solvent-free one-pot synthesis method for supported metal catalysts: High kinetic energies deposit metal particles on catalyst support particles, and in some cases also phase transformations may be facilitated. In this work we synthetized LOHC dehydrogenation catalysts by ball milling and studied the effect of catalyst synthesis method to its properties and performance.

Catalyst BM Pt/TiO₂ was synthetised by ball milling. The jars were filled with platinum and TiO₂ (anatase) powders, sealed, and milled for 3 h at 600 rpm. Catalyst IWI Pt/TiO₂ was synthetised by impregnating Pt to TiO₂ support material via incipient wetness technique. Both catalysts had 1 wt.% Pt loading. Continuous LOHC dehydrogenation experiments were operated in atmospheric pressure in a packed bed quarts glass reactor. The catalyst was activated by reduction. Methylcyclohexane was co-fed with nitrogen, and the reaction took place in vapor state (reactor bed 345–365 °C).

BM synthetized Pt/TiO₂ catalyst had a stable methylcyclohexane (MCH) conversion and 100% selectivity to toluene at 365 °C reactor temperature, whereas its IWI synthetized counterpart deactivated during the 15-hour-experiment. In lower temperature (345 °C) the two catalysts had nearly the same conversion at the end of the experiment.

As a result, a solvent-free, well reproducible, and scalable ball milling method for Pt-based LOHC dehydrogenation catalyst synthesis can be introduced.

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